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Atomic Structure Review Part 1 Review of Basic Atomic Structure Atomic Structure and Subatomic Particles **The Nucleus: Crash Course**

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Basic Atomic Structure Worksheet ANSWERS.

advertisement. Basic Atomic Structure

Worksheet ANSWERS 1 a) protons b) neutrons c)

electrons a) Positive b) Neutral c) negative

2 atomic number (or identity); charge 3

protons; electrons (in a neutral charge atom

only!); same 4 average atomic weight; mass 5

mass number; nucleus 6 neutrons; protons (or

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atomic number); mass number 7 Lithium = Li = 3 Iron = Fe=26 Oxygen = O = 8 Krypton = Kr = 36 8 (A neutral atom by definition has NO charge!

~~Basic Atomic Structure Worksheet ANSWERS~~

Basic Atomic Structure Worksheet H and the 1. 2. 4. 5. 6. 8. 9. The 3 particles of the atom are: protons (p) (e) Their respective charges are: The number of protons in one atom of an element determines the atom's number Of electrons determines the of the elemen. in one atom of an element.

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~~090412 Atomic Structure Worksheet 1~~

Basic Atomic Structure Worksheet ANSWERS 1 a) protons b) neutrons c) electrons a) Positive b) Neutral c) negative 2 atomic number (or identity); charge 3 protons; electrons (in a neutral charge atom only!); same 4 average atomic weight; mass 5 mass number; nucleus 6 neutrons; protons (or atomic number); mass number 7 Lithium = Li = 3 Bromine = Br = 35 Iron = Fe = 26 Copper = Cu = 29 Oxygen = O = 8 Mercury = Hg = 80 Krypton = Kr = 36 Helium = He = 2 8 (A neutral atom by definition has NO charge!

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Write two algebraic equations relating the number of protons and neutrons in an atom to that atom's atomic mass and atomic number:
Where, P = Number of protons. N = Number of neutrons. A = Atomic mass. Z = Atomic number

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2. 3. 4. 5. 6. 8. 9. The 3 particles of the atom are: Their respective charges are: The number of protons in one atom of an element determines the atom's number of electrons determines the of the element. The atomic number tells you the number of in one atom of an element. It also tells you the number of

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10. Name the element which has the following numbers of particles: a. b. c. d. 26 electrons, 29 neutrons, 26 protons 53 protons, 74 neutrons 2 electrons (neutral atoms)

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Atomic Basics Answer Key Part A: Atomic Structure 1. Draw five protons in the nucleus of the atom. Label them with their charge. 2.

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Draw six neutrons in the nucleus of the atom.
3. Draw two electrons in the first energy level and label them with their charge. 4. Draw three electrons in the second energy level and label them with their charge. 5.

~~Chemistry of Matter~~

Basic atomic structure worksheet h and the 1. Chlorine 35 42 atomic number equals the number of or atomic mass equals the number of identify the each of the parts of the box. Their respective charges are.

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~~worksheet~~

3. The atomic number tells you the number of _____ in one atom of an element. It also tells you the number of _____ in a neutral atom of that element. The atomic number gives the “identity” of an element as well as its location on the periodic table. No two different elements will

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Atomic Number, Mass Number, Symbols, Isotopes, Naming Isotopes, Bohr Model Drawing, Atomic Math Challenge, and questions like Label the parts of an atom on the diagram below. Atomic Structure Quiz, Electron Configuration, Electron Orbital Diagrams, ...

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9.1095×10^{-31} kg. Unit electrical charge is 1.6022×10^{-19} coulomb (C). The nucleus at the center of the atom contains one or more positively charged protons. All atoms of

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a given element have the same number of protons, which defines the element's atomic number, given the symbol Z.

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Atoms have 3 parts: protons, neutrons and
electrons. The majority of an atom is empty

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space. The nucleus contains the majority of an atom's mass and is incredibly small. If the atom was a size of a sports arena the nucleus would be as small as a pea! The hydrogen atom is the most abundant type of atom in the universe.

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