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chemical equations

class 10 chemistry

How To Balance

Chemical Equations

Chemical Reactions

and Equations

Balancing Chemical

Equations *Balancing*

Chemical Equations

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The first step to balance the equation is to write down the chemical formula of reactants that are listed on the left side of the chemical equation. After this, you can list down the products on the right hand side of the chemical equation.

There is an arrow

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between the sides,
signaling the direction
the reaction is
happening in.

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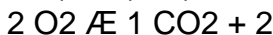
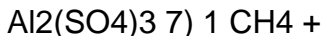
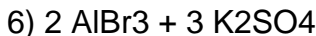
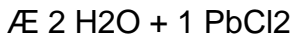
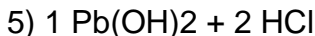
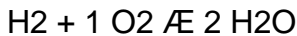
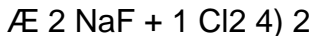
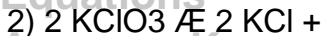
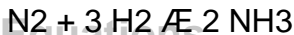
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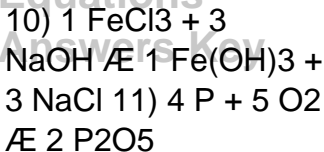
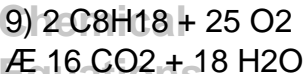
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Chemical Equations
Answers Key

equations below: 1) 1



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Balancing Chemical Equations

To set up an equation
in the Chemical
Equations Gizmo,
type the chemical
formulas into the text

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boxes of the Gizmo.

First, type in

“H₂+O₂” in the

Reactants box and

“H₂O” in the

Products box. This...

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A chemical equation

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is balanced by
changing the
subscripts YELLOW
coefficients LIGHT
BLUE anything
necessary DARK
GREEN 10 What is
the mass of Na?
 $2\text{NaN}_3 \rightarrow 2\text{Na} + 3\text{N}_2$
(500g) (?) (323g) 100
grams ORANGE 177
grams PINK 323
grams YELLOW

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the coefficients and
balance the following
equation. Remember
a coefficient cannot
be zero.

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*How To Balance
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Balancing an equation

If you just write an equation replacing names with formulae, it may not be balanced. The numbers of atoms of each element on the left must be the same as they are on the...

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Anne Marie
Helmenstine, Ph.D.
Updated January 04,
2019. A balanced
chemical equation
gives the number and
type of atoms

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participating in a reaction, the reactants, products, and direction of the reaction. Balancing an unbalanced equation is mostly a matter of making certain mass and charge are balanced on the reactants and products side of the reaction arrow.

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How to Balance Equations - Printable Worksheets

The must find the equation that corresponds to what is inside each atom and then follow the arrow with the correct answer to the next atom. Once an atom is used it cannot be revisited, even if an arrow with the correct

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answer points back to it, the student must find another path and make there way to the end.

*Science Maze
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of ...*

Balancing Equations
About Chemistry [http://
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Balance the following

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Chemical equations.

1. $\text{Fe} + 2 + 3 \cdot \text{H}_2\text{SO}_4$
4. $1 \cdot 2(\text{SO}_4) + 3 + 3$
- $\cdot \text{H}_2 \cdot 2 \cdot \text{C} \cdot 2 \cdot \text{H}_2 + 62 +$
- $7 \cdot \text{O}_2 \cdot 6 \cdot 2\text{O} \cdot \text{H} + 4$
- $\cdot \text{CO}_2 \cdot 3 \cdot \text{KOH} \cdot 3 +$
- $1 \cdot \text{H}_3\text{PO}_4 \cdot 1 \cdot 3\text{PO}_4$
- $4\text{K} + 3 \cdot \text{H}_2\text{O} \cdot 4.$
- $\text{SnO}_2 + 2 \cdot \text{H}_2 \cdot 1 \cdot$
- $\text{Sn} + 2 \cdot \text{H}_2\text{O} \cdot 5 \cdot \text{NH}_3$
- $4 + 5 \cdot \text{O}_2 \cdot 4 \cdot \text{NO}$
- $+ 6 \cdot \text{H}_2\text{O} \cdot 6 \cdot \text{KNO}_3$
- $2 + 1 \cdot \text{H}_2\text{CO}_3 \cdot 1 \cdot$
- $2\text{CO}_3 + 2 \cdot \text{HNO}_3 \cdot 3.$

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7. B. 2. Br. 61 + 6 .
HNO 3. 2. B(NO 3) 3
+ 6 . HBr . 8. BF. 3. 2
+ 3 . Li. 2SO 3. 1.
2(SOB 3) 3 + 6

Name: Date:

Balancing Equations

Balancing Equations:

Answers to Practice

Problems. 1.

Balanced equations.

(Coefficients equal to

one (1) do not need to

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be shown in your answers). (a) $2 \text{Fe} + 3 \text{Cl}_2 \rightarrow 2 \text{FeCl}_3$. (b) $4 \text{Fe} + 3 \text{O}_2 \rightarrow 2 \text{Fe}_2\text{O}_3$. (c) $2 \text{Fe} + 3 \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3 \text{H}_2$.

Chemical Equations Answers - Teacher Worksheets

There are 10
chemical equations
for students to

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Chemical
Equations
Answers Key

balance. Each equation comes with a helpful chart so students practice counting up the number of atoms on each side and making sure they match. This worksheet is perfect for a student learning how to balance an equation for the first time! Give as classwork, assign as

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homework, or leave
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